

B.Sc. I

Paper II Organic Chemistry

**ELIMINATION REACTIONS – E1, E2
REARRANGEMENT REACTION**

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ELIMINATION REACTION

An elimination reaction is a type of organic reaction in which two substituents are removed from a molecule in either a one-step or two-step mechanism.

The one-step mechanism is known as the **E2 reaction**, and the two-step mechanism is known as the **E1 reaction**.

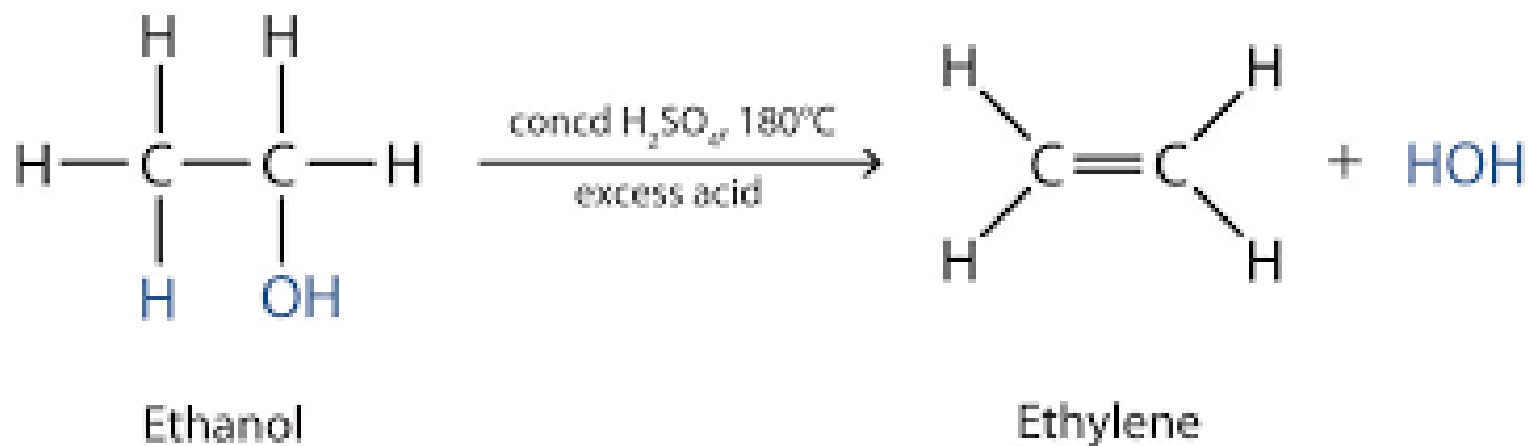
E2 is bimolecular (second-order) while E1 is unimolecular (first-order).

In most organic elimination reactions, at least one hydrogen atom (H^+) is lost, to form the double bond.

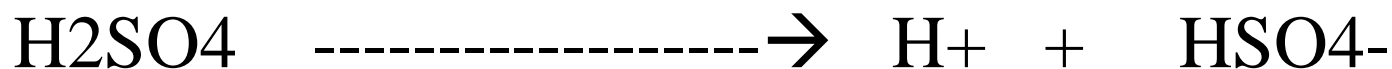
When the substrate is **asymmetric**, regioselectivity is determined by **Saytzeff's rule** or through Hofmann elimination.

1. Dehydration of Alcohols
2. Dehydrohalogenation of Alkyl halides

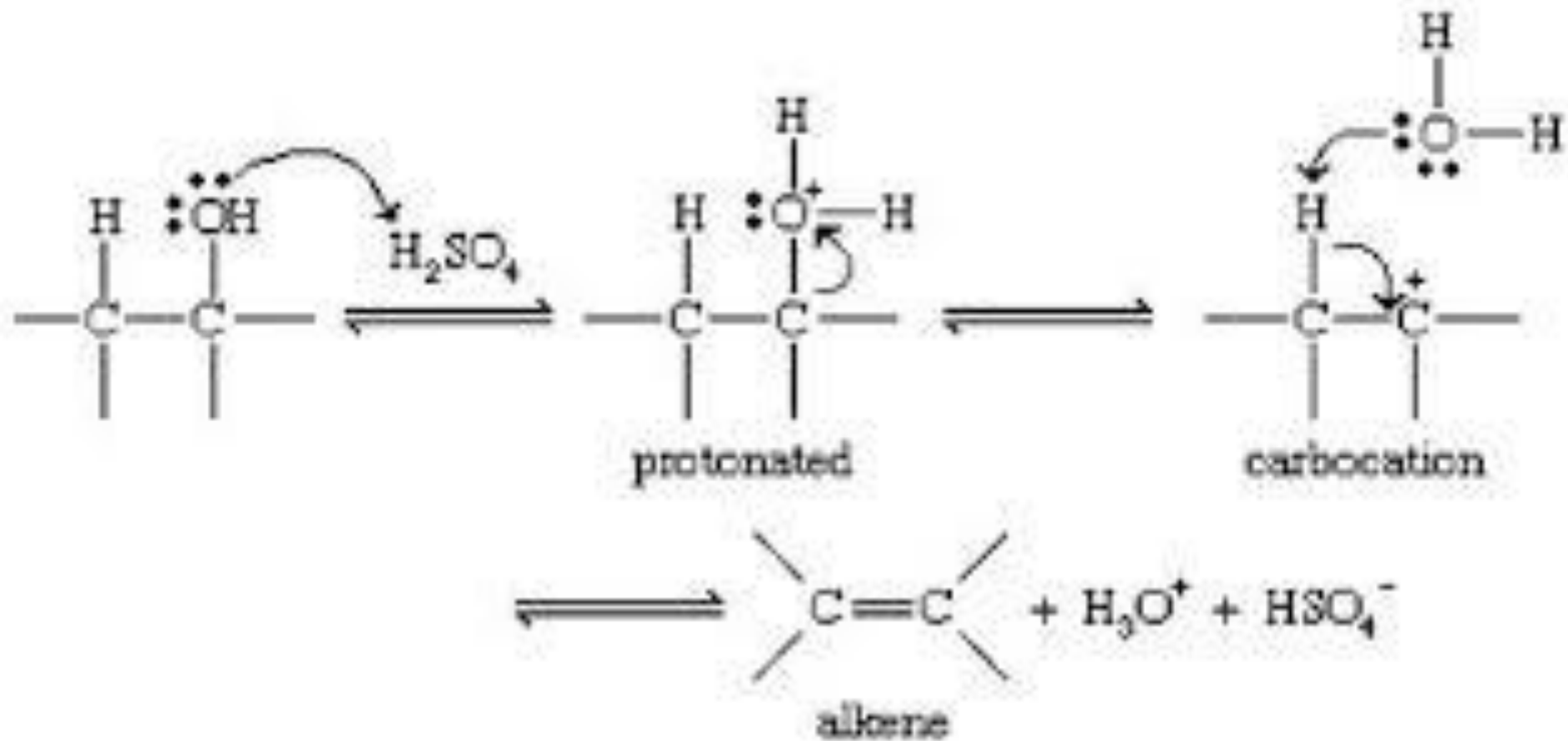
DEHYDRATION OF ALCOHOLS



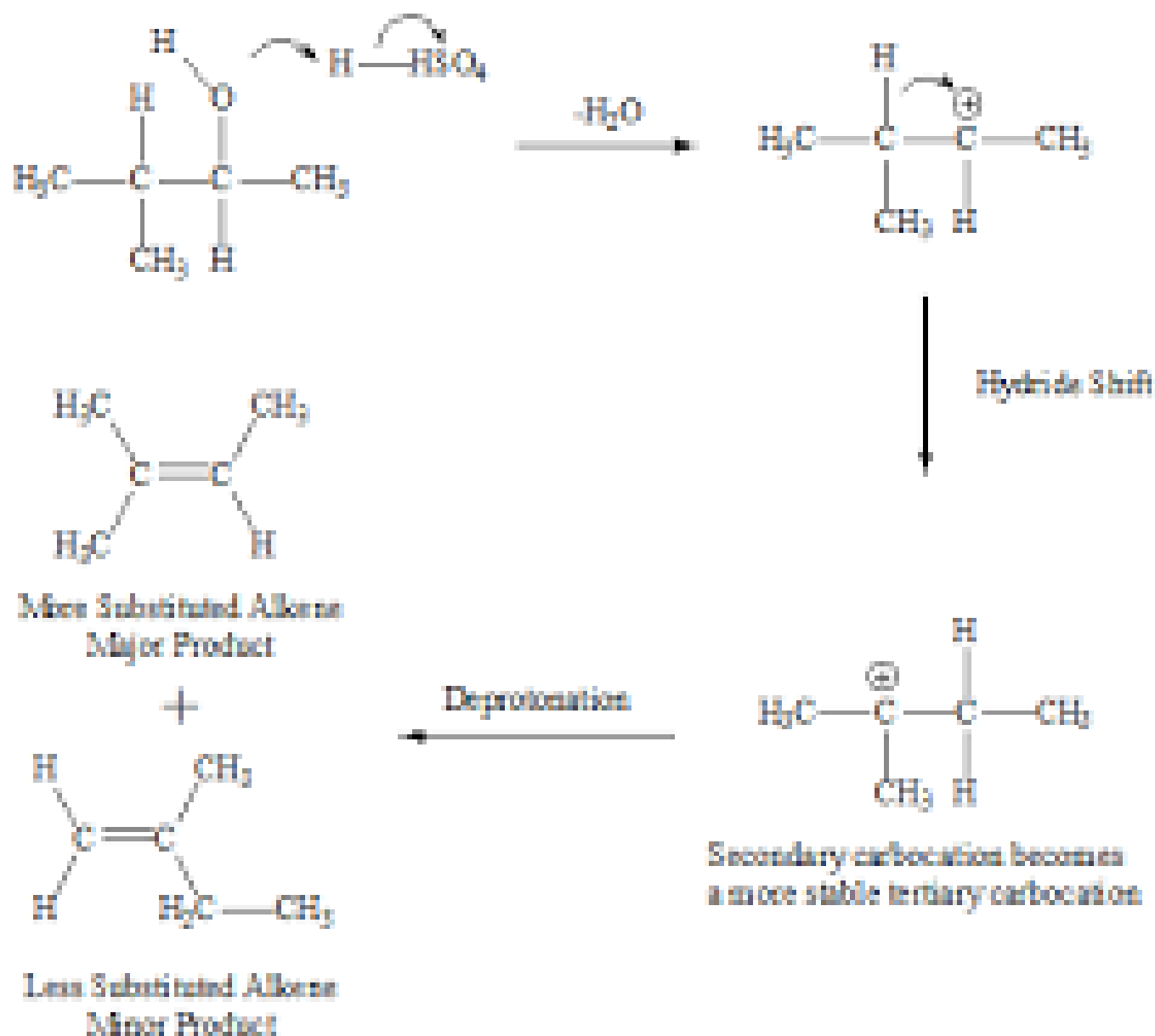
CONCENTRATED SULPHURIC ACID IS THE DEHYDRATING AGENT
DEHYDRATION IS CARRIED OUT IN PRESENCE OF CONCENTRATED H₂SO₄



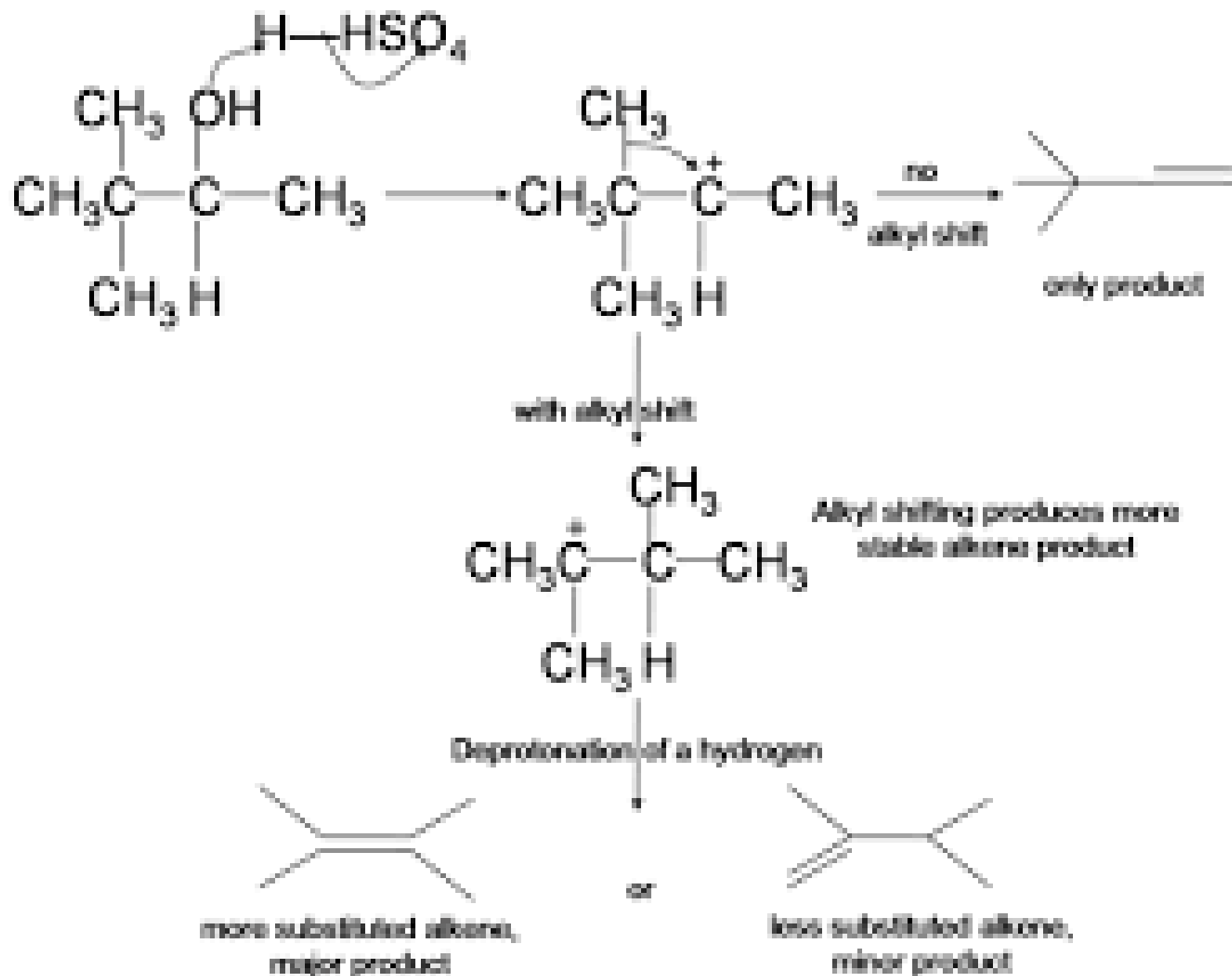
DEHYDRATION OF ALCOHOLS



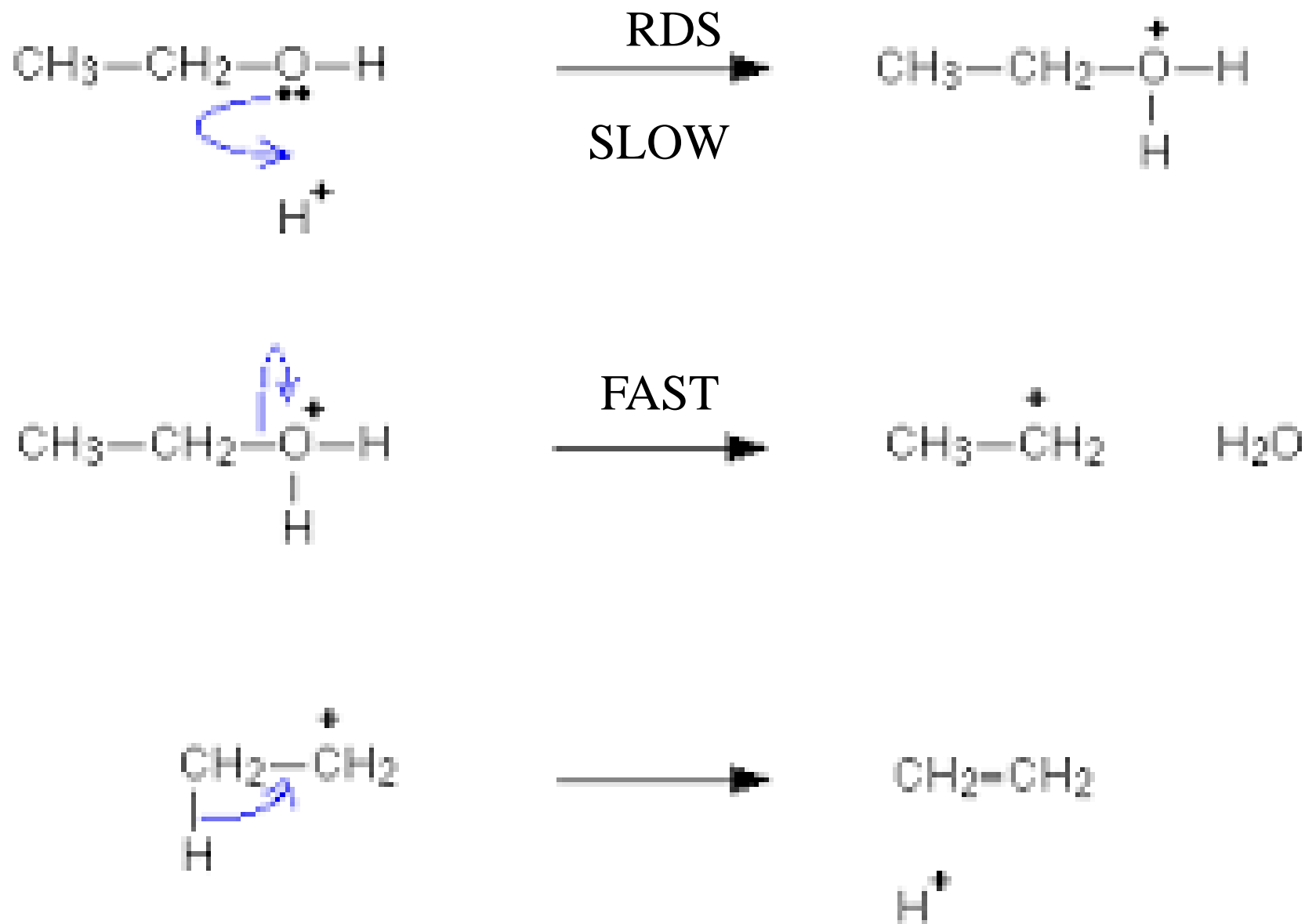
DEHYDRATION OF ALCOHOLS



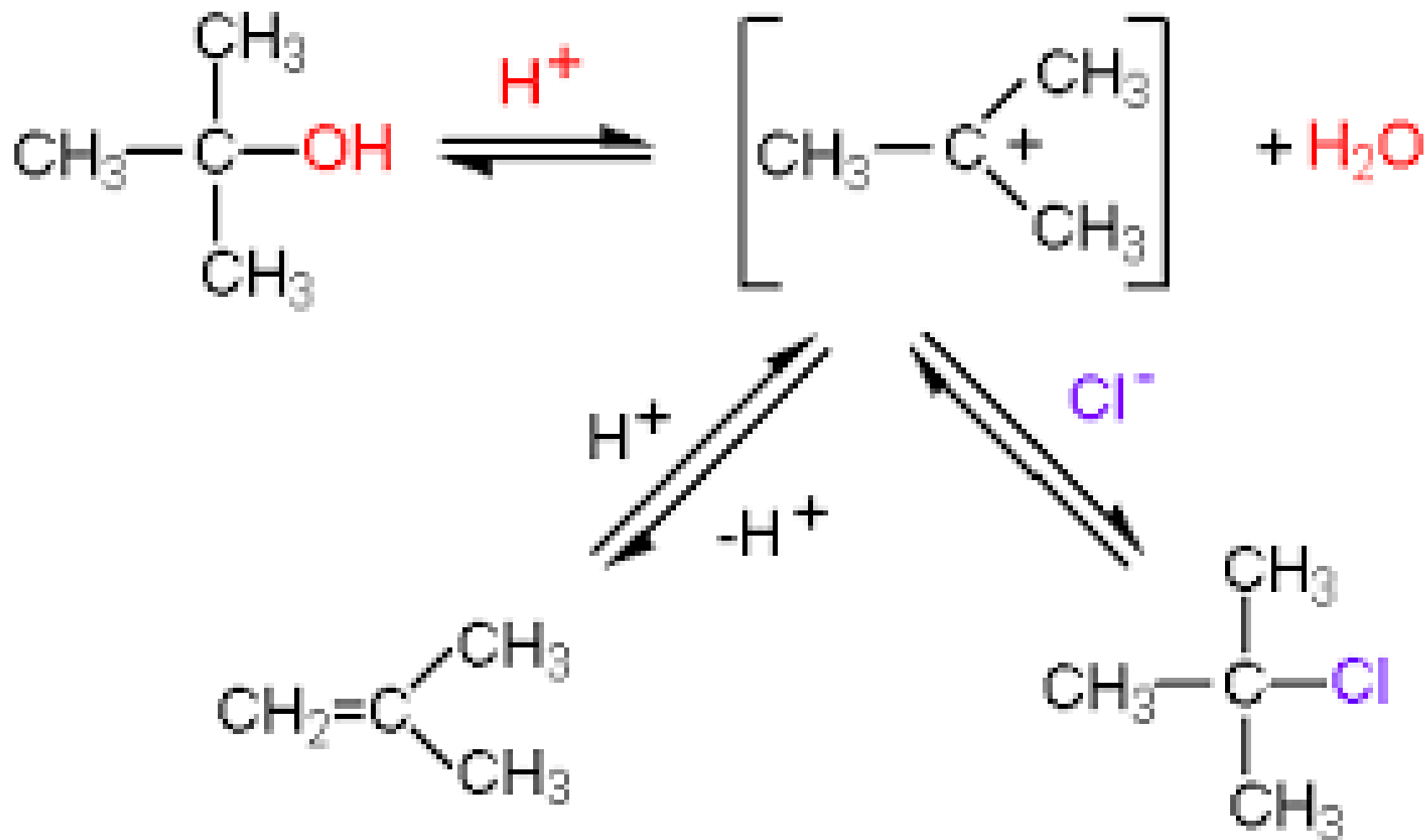
DEHYDRATION OF ALCOHOLS



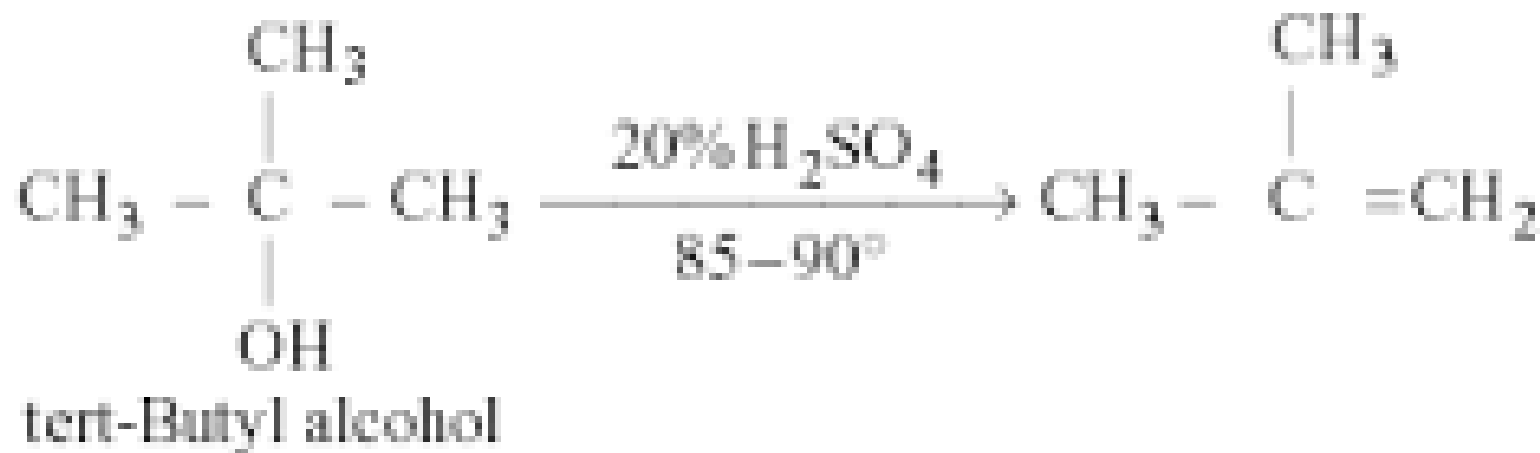
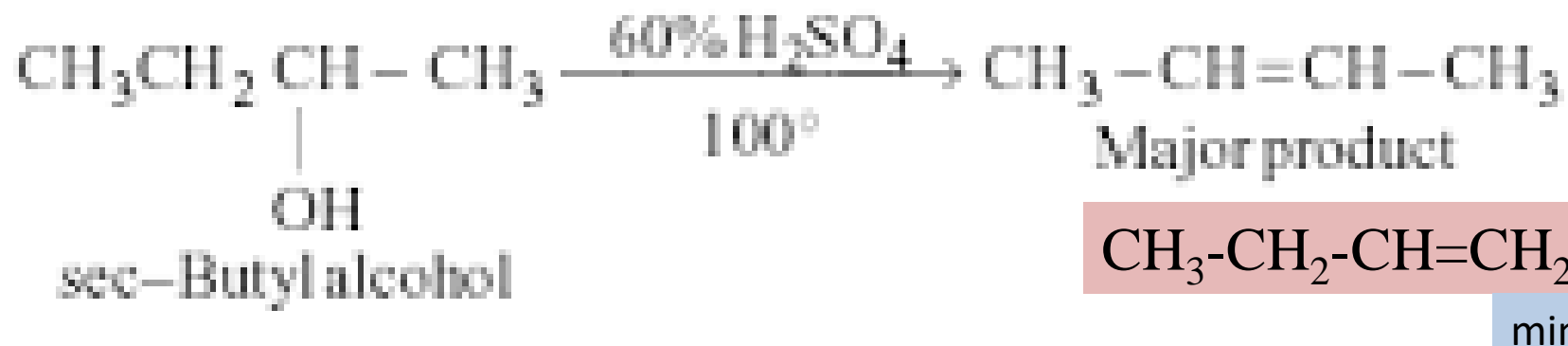
E2 – BIMOLECULAR ELIMINATION



E1 – UNIMOLECULAR ELIMINATION



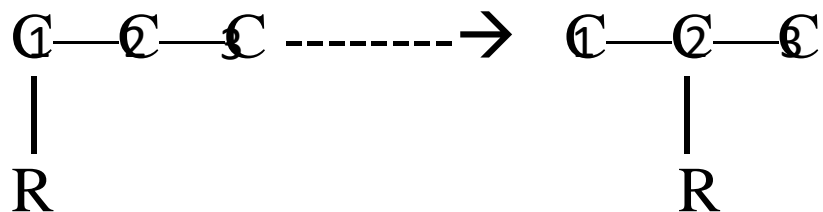
DEHYDRATION OF ALCOHOLS



REARRANGEMENT REACTION

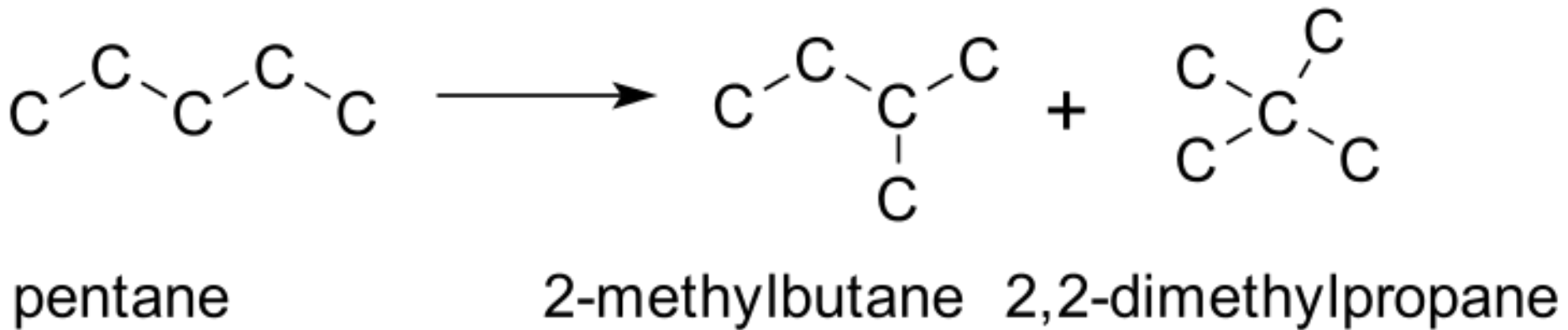
A rearrangement reaction is a broad class of organic reactions where the carbon skeleton of a molecule is rearranged to give a structural isomer of the original molecule. Often a substituent moves from one atom to another atom in the same molecule.

In the example below the substituent R moves from carbon atom 1 to carbon atom 2.

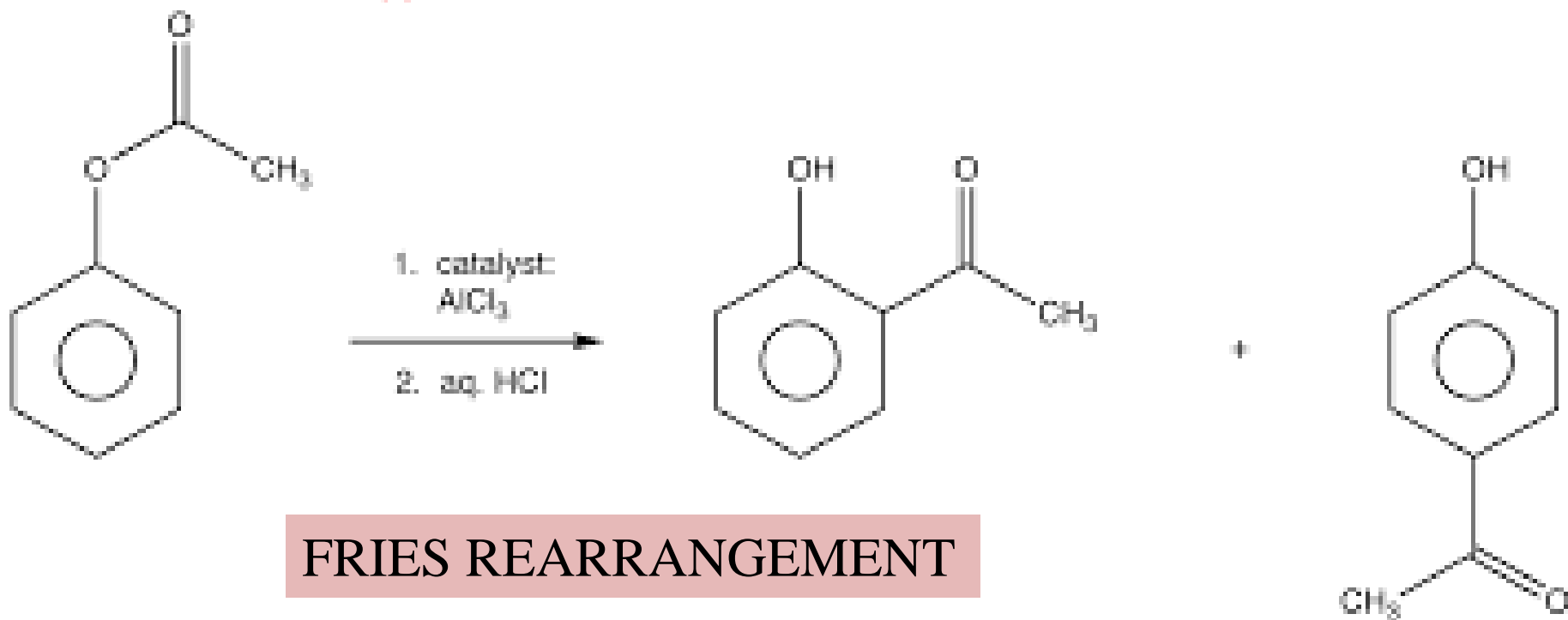
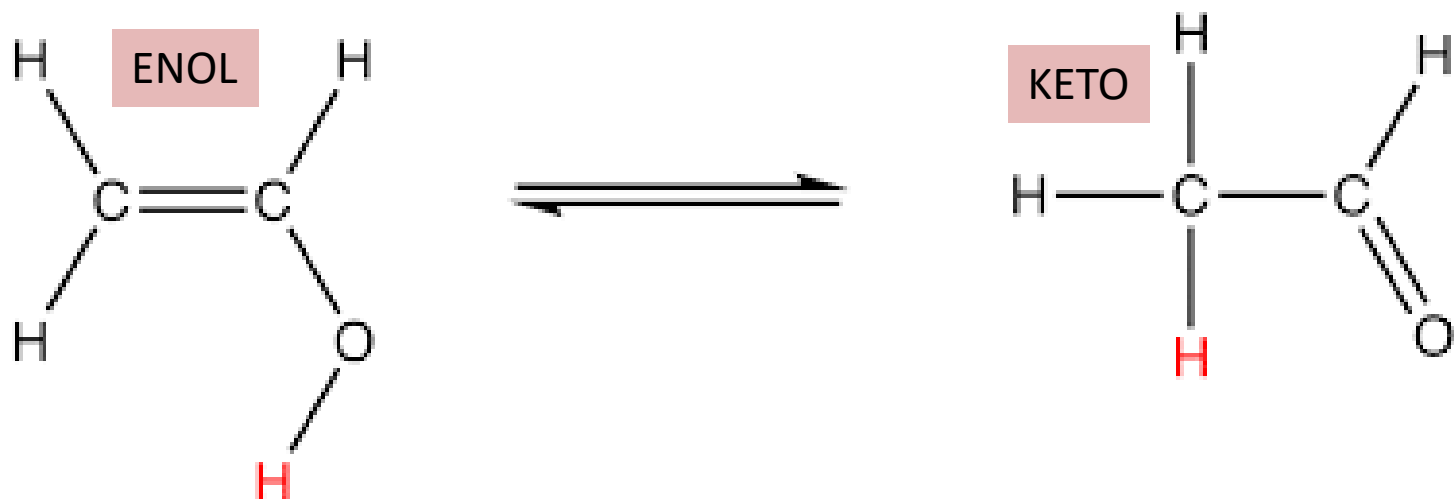


REARRANGEMENT REACTION

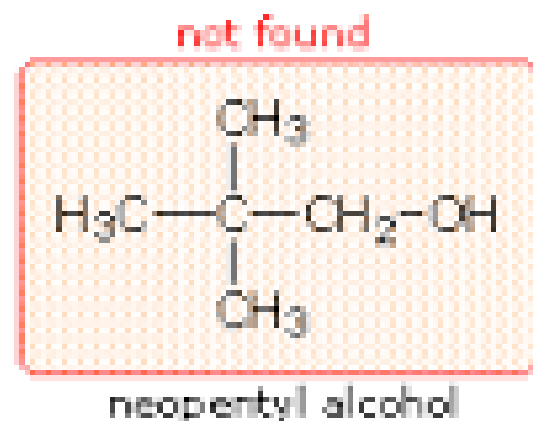
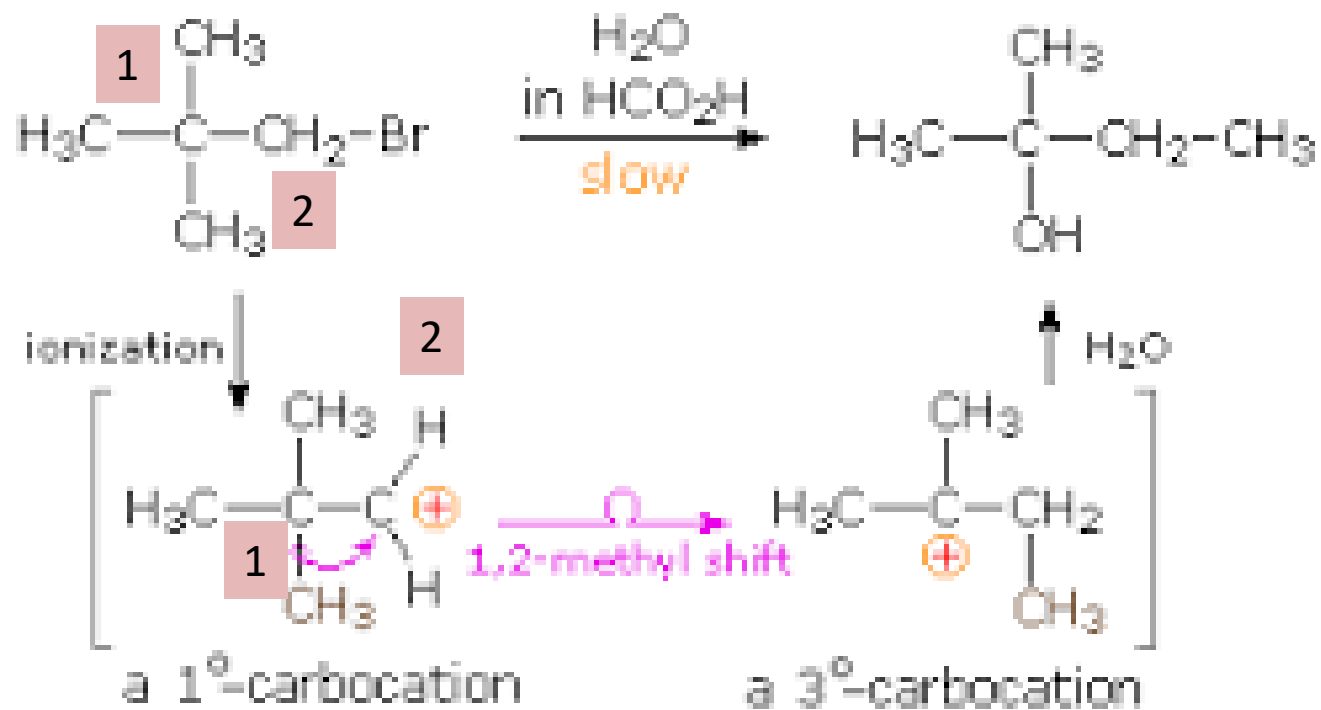
In general straight-chain alkanes, are converted to branched isomers by heating in the presence of a catalyst. Examples include isomerisation of n-butane to isobutane and pentane to isopentane.



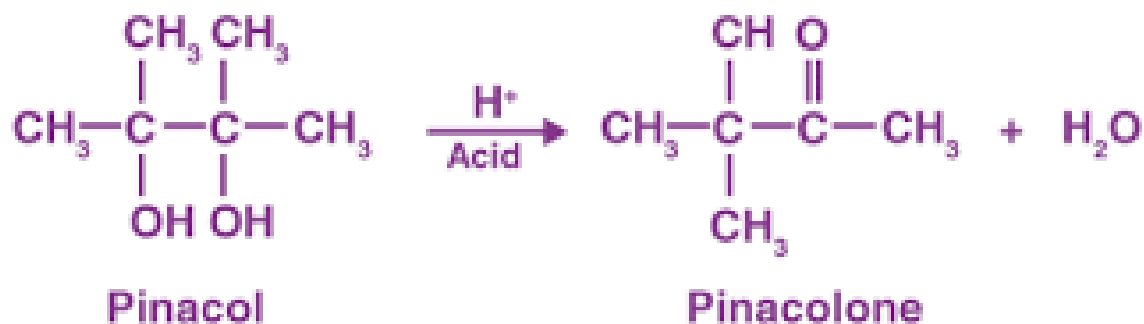
REARRANGEMENT REACTIONS



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